

# ENTHONE

## MATERIAL SAFETY DATA SHEET

Health	3
Flammability	0
Reactivity	1
Personal Protection	

### In Case of Emergency

CHEMTREC Number  
(800) 424-9300

## Section 1. Chemical Product and Company Identification

**Product Name** MICROFAB® CU-300 ADDITIVE  
**Product Code Number(s)** 250816  
250817-001  
**Material Uses** Specialty chemicals for the electronics and surface finishing industries.  
**Manufacturer** ENTHONE  
350 Frontage Road  
West Haven, CT 06516  
(203) 799-4917  
(203) 799-8179 (fax)  
www.cooksonelectronics.com  
**Supersedes Date** NEW **Revision No.** 1  
**Print Date** 2/11/2003. **Validation Date** 2/11/2003.  
**Prepared by** Anton Mayer - Regulatory Specialist

## Section 2. Composition, Information on Ingredients

Name	CAS #	% by Weight
SULFURIC ACID	7664-93-9	5-10
NON-HAZARDOUS ORGANIC ACID SALT	Proprietary	<1
FORMALDEHYDE	50-00-0	<0.1
WATER	7732-18-5	90-100

*All ingredients comply with applicable rules or orders under TSCA*

## Section 3. Hazards Identification

**Physical State and Appearance** Liquid. **Odor** Acidic. **Color** Colorless.

### Emergency Overview

DANGER!

May be fatal if swallowed. May cause burns to mouth, throat and stomach.

### Routes of Entry

Dermal contact. Eye contact. Inhalation. Ingestion.

### Potential Acute Health Effects

**Eyes** Very hazardous in case of eye contact (corrosive). Inflammation of the eye is characterized by redness, watering, and itching.

**Skin** Very hazardous in case of skin contact (corrosive). Skin contact may produce burns. Skin inflammation is characterized by itching, scaling, reddening, or, occasionally, blistering.

**Inhalation** Very hazardous in case of inhalation (lung corrosive). Inhalation of the spray mist may produce severe irritation of respiratory tract, characterized by coughing, choking, or shortness of breath.

**Ingestion** May be fatal if swallowed. May cause burns to mouth, throat and stomach.

### Medical Conditions Caused or Aggravated by Overexposure

Prolonged contact with skin, eyes, or respiratory tract may result in burns or tissue damage due to corrosive effects. Repeated or prolonged skin contact can cause drying and cracking of the skin (dermatitis).

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## Section 4. First Aid Measures

<b>Eye Contact</b>	Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Cold water may be used. Get medical attention immediately.
<b>Skin Contact</b>	In case of contact, immediately flush skin with plenty of water for at least 15 minutes while removing contaminated clothing and shoes. Get medical attention immediately.
<b>Inhalation</b>	If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention immediately.
<b>Ingestion</b>	If swallowed, do not induce vomiting unless directed to do so by medical personnel. Never give anything by mouth to an unconscious person. Loosen tight clothing such as a collar, tie, belt or waistband. Get medical attention immediately.

## Section 5. Fire Fighting Measures

<b>Flammability of the Product</b>	Non-flammable.
<b>Flash Points</b>	Not applicable.
<b>Products of Combustion</b>	This material is not combustible, however in a fire it may give off sulfur oxides (SO <sub>2</sub> , SO <sub>3</sub> ...)
<b>Fire Fighting Media and Instructions</b>	Use DRY chemicals, CO <sub>2</sub> , water spray or foam.
<b>Protective Equipment (Fire)</b>	Wear NIOSH approved self-contained breathing apparatus or equivalent and full protective gear.
<b>Unusual Firefighting Hazards</b>	May react with amphoteric metals (such as aluminum, zinc, tin) generating hydrogen gas which will burn or explode if ignited.

## Section 6. Accidental Release Measures

<b>Small Spill and Leak</b>	Absorb with an inert material and place in an appropriate waste disposal container. If necessary: <b>Neutralize the residue with a dilute solution of sodium carbonate.</b>
<b>Large Spill and Leak</b>	Stop leak if without risk. Absorb with DRY earth, sand or other non-combustible material. Do not touch spilled material. Prevent entry into sewers, basements or confined areas; dike if needed. Call for assistance on disposal. <b>Neutralize the residue with a dilute solution of sodium carbonate.</b>

## Section 7. Handling and Storage

<b>Handling</b>	Do not ingest. Do not breathe vapor or mist. Keep container closed. Use only with adequate ventilation. Wash thoroughly after handling.
<b>Storage</b>	Keep container tightly closed. Keep container in a cool, well-ventilated area.

## Section 8. Exposure Controls, Personal Protection

<b>Engineering Controls</b>	Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapors below their respective occupational exposure limits. Ensure that eyewash stations and safety showers are proximal to the work-station location.
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### Personal Protection

*Eyes* Face shield. Splash goggles.

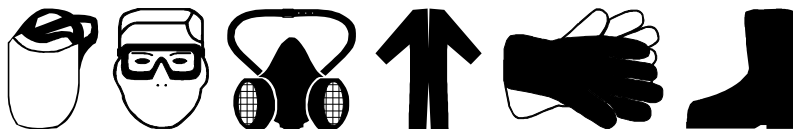
*Body* Chemical resistant protective suit.

*Respiratory* Be sure to use a MSHA/NIOSH approved respirator or equivalent. Respirator selection must be based on known or anticipated exposure levels, the hazards of the product, and the safe working limits of the selected respirator.

*Hands* Butyl rubber gloves. Neoprene gloves.

*Feet* Boots.

### Protective Equipment (Pictograms)



<b>Personal Protection in Case of a Large Spill</b>	Face shield. Splash goggles. Full suit. Boots. Gloves. A self-contained breathing apparatus should be used to avoid inhalation of the product. Suggested protective clothing might not be sufficient; consult a specialist BEFORE handling this product.
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### Ingredient Name

### Exposure Limits

SULFURIC ACID

**ACGIH (United States).**  
TWA: 0.25 ppm 8 hour(s).  
**OSHA (United States).**  
TWA: 0.25 ppm 8 hour(s).  
**ACGIH (United States).**  
STEL: 0.75 ppm 15 minute(s).  
**OSHA (United States).**  
STEL: 0.75 ppm 15 minute(s).  
**ACGIH TLV (United States, 2000).**  
STEL: 3 mg/m<sup>3</sup> 15 minute(s).  
TWA: 1 mg/m<sup>3</sup> 8 hour(s).  
**NIOSH REL (United States, 2000).**  
TWA: 1 mg/m<sup>3</sup> 10 hour(s).  
**OSHA PEL 1989 (United States, 1989).**  
TWA: 1 mg/m<sup>3</sup> 8 hour(s).

Consult local authorities for acceptable exposure limits.

## Section 9. Physical and Chemical Properties

<b>Physical State and Appearance</b>	Liquid.	<b>Odor</b>	Acidic.	<b>Color</b>	Colorless.
<b>pH</b>	<1 [Acidic.]				
<b>Boiling Point</b>	100 °C				
<b>Melting/Freezing Point</b>	0 °C				
<b>Specific Gravity</b>	1.054 (Water = 1)				
<b>VOC</b>	Not applicable.				
<b>Solubility</b>	Soluble in cold water, hot water.				

## Section 10. Stability and Reactivity

<b>Stability and Reactivity</b>	Stable under recommended storage and handling conditions (see section 7).
<b>Incompatible Substances</b>	Reactive with oxidizing agents, alkalis. Cyanides. Hydrogen, a highly flammable and explosive gas, may be generated by the action of acids on many metals.
<b>Hazardous Decomposition Products</b>	In a fire: toxic sulfur oxides.
<b>Hazardous Polymerization</b>	Will not occur.

## Section 11. Toxicological Information

### Toxicity Data

<u>Ingredient Name</u>	<u>Test</u>	<u>Result</u>	<u>Route</u>	<u>Species</u>
SULFURIC ACID	LD50	2140 mg/kg	Oral	Rat
	LC50	347 ppm (1 hours)	INHALATION	Rat

**Chronic Effects on Humans** **CARCINOGENIC EFFECTS:** Classified A2 (Suspected for human.) by ACGIH [SULFURIC ACID].  
**MUTAGENIC EFFECTS:** None identified.  
**TERATOGENIC EFFECTS:** None identified.

**Special Remarks on Toxicity** Inhalation of the spray mist may produce severe irritation of respiratory tract, characterized by coughing, choking, or shortness of breath. This product contains less than 0.1% formaldehyde. Formaldehyde may cause sensitization and reduced vision. Lifetime study of rats exposed to formaldehyde gas indicated development of nasal cancers.

Any component listed in this section that is not listed in Section 2 is present in the product in concentrations below legal disclosure limits ( 1% for hazardous components and 0.1% for carcinogens )

## Section 12. Ecological Information

Enthone has not conducted specific studies on the ecotoxicity or environmental fate of this product.

## Section 13. Disposal Considerations

**Waste Information** Waste must be disposed of in accordance with federal, state and local environmental control regulations.

## Section 14. Transport Information

**DOT Classification** 8  
**UN number** UN2796  
**Proper shipping name** SULFURIC ACID, SOLUTION  
**Packing Group** II

*This Transport Information applies only to the Product Code Number(s) listed in Section 1. Other container sizes may require different Transport Information. If assistance is required, contact Regulatory Affairs at 203-799-4936.*

## Section 15. Regulatory Information

**U.S. Federal Regulations** SARA 313 toxic chemical notification and release reporting: SULFURIC ACID

*All ingredients comply with applicable rules or orders under TSCA*

**State Regulations**

California prop. 65: None identified.

*Any component listed in this section that is not listed in Section 2 is present in the product in concentrations below legal disclosure limits (1% for hazardous components and 0.1% for carcinogens).*

## Section 16. Other Information

**Definition of Terms**

ACGIH	American Conference of Governmental Industrial Hygienists
Ceiling	Maximum exposure limit defined by OSHA
CAS	Chemical Abstract Service
IARC	International Agency for Research on Cancer
NIOSH	National Institute for Occupational Safety and Health
NTP	National Toxicology Program
OSHA	Occupational Safety and Health Administration
PEL	Permissible Exposure Limit
REL	Recommended Exposure Limit
RTK	Right to Know
SARA	Superfund Amendments and Reauthorization Act
STEL	Short Term Exposure Limit
TLV	ACGIH Threshold Limit Value
TLV-C	ACGIH Threshold Limit Value, Ceiling
TRADE SECRET	Claimed as allowed under 29CFR§1910.1200
TSCA	Toxic Substances Control Act

### Disclaimer

*This Material Safety Data Sheet may be used to comply with OSHA's Hazard Communication Standard, 29CFR§1910.1200. Enthone furnishes the data contained herein in good faith at customer's request without liability or legal responsibility for same whatsoever, and no warranty or guarantee, express or implied, is made with respect to such data; nor does Enthone grant permission, recommendation, or inducement to infringe any patent whether owned by Enthone or others. The data is offered solely for your information and consideration. Since conditions of use are beyond Enthone's control, user assumes all responsibility and risk.*

